



Journal of Molecular Catalysis A: Chemical 235 (2005) v-xiv

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## Contents

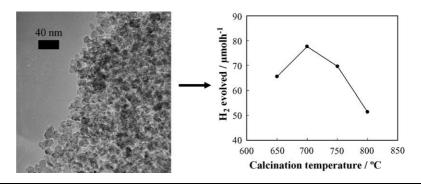
## Articles

(2005) 1

## Thammanoon Sreethawong, Supachai Ngamsinlapasathian, Yoshikazu Suzuki, Susumu Yoshikawa

Journal of Molecular Catalysis A: Chemical 235

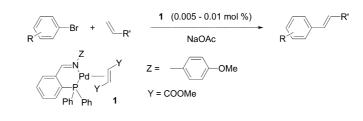
Nanocrystalline mesoporous Ta<sub>2</sub>O<sub>5</sub>-based photocatalysts prepared by surfactant-assisted templating sol-gel process for photocatalytic H<sub>2</sub> evolution Typical TEM image of nanocrystalline mesoporous  $Ta_2O_5$  prepared by surfactant-assisted templating sol-gel process of laurylamine hydrochloride/tantalum pentaethoxide modified with acetylacetone system and calcined at 700 °C for 4 h, exhibiting high photocatalytic H<sub>2</sub> evolution activity.



Alberto Scrivanti, Matteo Bertoldini, Ugo Matteoli, Valentina Beghetto, Simonetta Antonaroli, Alessia Marini, Bruno Crociani

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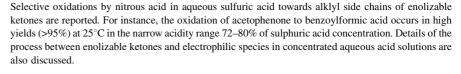
Highly efficient Heck olefin arylation in the presence of iminophosphine–palladium(0) complexes

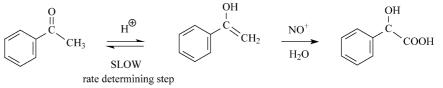


# N.C. Marziano, L. Ronchin, C. Tortato, A. Zingales, L. Scantamburlo

Journal of Molecular Catalysis A: Chemical 235 (2005) 17

Selective oxidations by nitrosating agents. Part 1: Oxidations of ketones in concentrated sulfuric acid

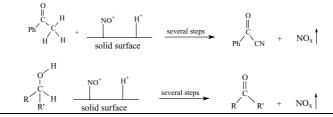




# N.C. Marziano, L. Ronchin, C. Tortato, S. Ronchin, A. Vavasori

Journal of Molecular Catalysis A: Chemical 235 (2005) 26

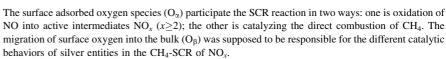
Selective oxidations by nitrosating agents. Part 2: Oxidations of alcohols and ketones over solid acid catalysts The reactivity of a nitrosating agent (N<sub>2</sub>O<sub>3</sub>) on oxidations of alcohols and of acetophenone were tested. Active electrophylic surface nitrosonium ions (NO<sup>+</sup>) was detected on H<sub>2</sub>SO<sub>4</sub>/SiO<sub>2</sub> catalysts by Raman spectroscopy, suggesting a surface ionic mechanism of oxidation. Alcohols are selectively oxidized to ketones and aldehydes in high yield, at 25°C in 1,2-dichlorethane and using sulfonated styrene divinyl benzene resins as catalysts. Under the same reaction conditions, acetophenone is selectively oxidized to benzoyl cyanide in high yield and selectivity. The comparison with oxidation carried out in aqueous solution and the reactivity of the intermediates isolated in such system gives new insight on the mechanism of the heterogeneous reactions.

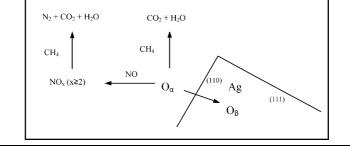


#### Chuan Shi, Mojie Cheng, Zhenping Qu, Xinhe Bao

Journal of Molecular Catalysis A: Chemical 235 (2005) 35

On the correlation between microstructural changes of Ag-H-ZSM-5 catalysts and their catalytic performances in the selective catalytic reduction of  $NO_x$  by methane



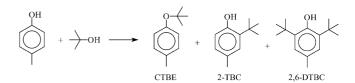


### S. Sarish, Biju M. Devassy, S.B. Halligudi

Journal of Molecular Catalysis A: Chemical 235 (2005) 44

*tert*-Butylation of *p*-cresol over WO<sub>x</sub>/ZrO<sub>2</sub> solid acid catalysts

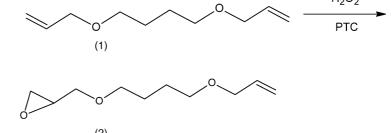
The alkylation of *p*-cresol with *tert*-butanol was investigated using WO<sub>x</sub>/ZrO<sub>2</sub> as catalysts. Under the optimised reaction conditions of 130°C, *tert*-butanol/phenol molar ratio 3 and WHSV  $4.3h^{-1}$ , the most active catalyst 15% WO<sub>3</sub>/ZrO<sub>2</sub> calcined at 800°C, gave 69.8% *p*-cresol conversion with selectivity to 2-*tert*-butyl-*p*-cresol 92.4%, 2,6-di-*tert*-butyl-*p*-cresol 6.3% and *p*-cresol *tert*-butyl ether 1.3%.



## E. Kaczmarczyk, E. Janus, E. Milchert

Journal of Molecular Catalysis A: Chemical 235 (2005) 52

Epoxidation of 1,4-diallyloxybutane to 1-allyloxy-4-glycidyloxybutane by the method of phase transfer catalysis The epoxidation of 1,4-diallyloxybutane (1) to 1-allyloxy-4-glycidyloxybutane (2) with 30 wt% hydrogen peroxide using phase transfer catalyst has been investigated.  $H_2O_2$ 

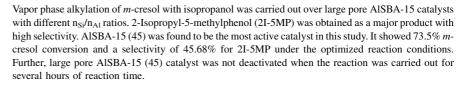


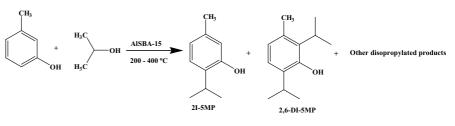
(2) Two different catalytic systems were used: phosphoric(V) acid - sodium tungsten(VI) dihydrate ( $PO_4^{3-}/WO_4^{2-}$ ) and phosphorotungstic acid hydrate  $H_3PW_{12}O_{40}$ . Methyltrioctylammonium chloride under trade name Aliquat<sup>®</sup>336 as phase transfer catalyst was used. The process in the presence of an auxiliary solvents was carried.

# A. Vinu, G. Satish Kumar, K. Ariga, V. Murugesan

Journal of Molecular Catalysis A: Chemical 235 (2005) 57

Preparation of highly ordered mesoporous AlSBA-15 and its application to isopropylation of *m*-cresol

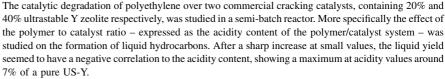


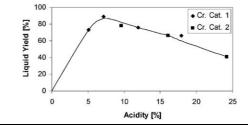


## Nnamso S. Akpanudoh, Karishma Gobin, George Manos

Journal of Molecular Catalysis A: Chemical 235 studied on t (2005) 67 seemed to h

Catalytic degradation of plastic waste to liquid fuel over commercial cracking catalysts. Effect of polymer to catalyst ratio/acidity content





Various ionic liquids enhanced free-solvent alkylations of  $\alpha$ -methylnaphthalene with long-chain olefins

have been investigated. It is found that the metal ion modified alkylpyridinium chloroaluminate ionic liquids (RPyBr-AlCl<sub>3</sub>), especially butylpyridinium chloroaluminate (BuPyBr-AlCl<sub>3</sub>, 0.71–0.75 molar

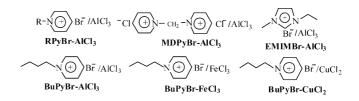
fraction of aluminum trichloride) indicated excellent catalytic performance for the alkylations of  $\alpha$ -

### Zhongkui Zhao, Bing Yuan, Weihong Qiao, Zongshi Li, Guiru Wang, Lübo Cheng

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methylnaphthalene.

The metal ion modified ionic liquids promoted free-solvent alkylations of  $\alpha$ -methylnaphthalene with long-chain olefins



Tetrameric ethoxo-bridged copper(II)-benzoylacetonate complex Cu4(bnac)4(µ-OEt)4 was isolated and

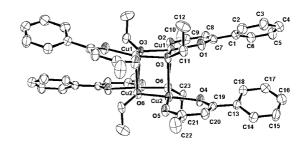
characterized. Single crystal X-ray structure of the tetrameric complex is also reported. The in situgenerated  $Cu_2(bnac)_2(\mu-OEt)_2(4R-py)_4$  complexes show catecholase 2 activity. The study reveals the

importance of the electron density on the copper centers influenced by the nature of axial ligands.

## József Kaizer, Róbert Csonka, Gábor Speier, Michel Giorgi, Marius Réglier

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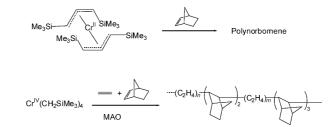
Synthesis, structure and catecholase-like activity of a new dicopper(II) complex with benzoylace-tonate ligand



# Timothy J. Woodman, Yann Sarazin, Shaun Garratt, Gerhard Fink, Manfred Bochmann

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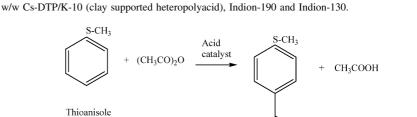
Chromium allyl and alkyl catalysts for the vinyl polymerization of norbornene and ethylene–norbornene copolymerizations The catalytic behavior of Cr(II) and Cr(IV) catalyst precursors for the polymerization of ethylene and norbornene are compared. Cr(allyl)<sub>2</sub>/MAO catalysts polymerize norbornene but not ethylene, whereas Cr(II) undergoes one-electron transfer with  $B(C_6F_5)_3$  to generate  $[Cr^{III}(allyl)_2]^+$  which readily polymerizes ethylene. By contrast, the Cr(IV) complex Cr(CH<sub>2</sub>SiMe<sub>3</sub>)<sub>4</sub>/MAO oligomerizes norbornene but is highly active for ethylene/norbornene copolymerizations to high molecular weight copolymers with up to 60mol% norbornene content. Significantly, activity increases with norbornene concentration.



#### Ganapati D. Yadav, Ramesh D. Bhagat

Journal of Molecular Catalysis A: Chemical 235 (2005) 98

Experimental and theoretical analysis of Friedel– Crafts acylation of thioanisole to 4-(methylthio)acetophenone using solid acids



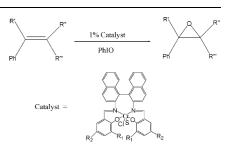
4-(Methylthio)-acetophenone

Experimental and theoretical analysis of selective acylation of thioanisole with acetic anhydride to 4-(methylthio)acetophenone was performed with Amberlyst-15, which was the best catalyst vis-a-vis 20%

Zheng-Kai Li, Lei Liang, Li Yang, Hua Chen, Xiang-Ge Zhou

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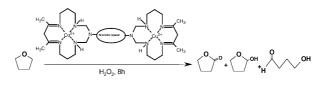
Asymmetric epoxidation catalyzed by Cr(III)binaphthyl Schiff base complexes A series of chromic complexes with binaphthyl Schiff base as ligands are synthesized. Their catalytic abilities in asymmetric epoxidation and the effects of reaction conditions such as temperature, solvent and additive have also been studied, leading to the highest ee of 65% for 4-chlorostyrene; the reaction seems to involve a Cr(V)-oxo active species.



## Masoud Salavati-Niasari, Ahmad Amiri

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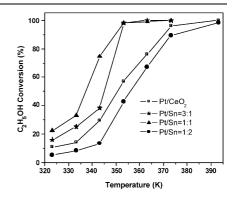
Binuclear copper(II) complexes of new bis(macrocyclic) 16-membered pentaaza subunits are linked together by bridging nitrogen of amine: Synthesis, characterization and catalytic activity New bis(macrocyclic) dicopper(II) complexes containing phenylene bridges between 16-membered pentaaza macrocyclic subunits have been synthesized via one-pot template condensation of aromatic nitrogen–nitrogen linker, formaldehyde, 1,3-diaminopropane, copper(II) and 2,4-pentanedione in a 1:4:4:2:2 molar ratio results in the formation of new series of binuclear copper(II) complexes ({[Cu(Me<sub>2</sub>[16]aneN<sub>5</sub>)]<sub>2</sub>R](ClO<sub>4</sub>)<sub>4</sub> and {[Cu(Me<sub>2</sub>[16]aneN<sub>5</sub>)]<sub>2</sub>RCl<sub>4</sub>}). Elemental analyses, IR, UV–vis spectroscopy, conductometric and magnetic measurements have been used to characterize the new bis(macrocyclic) binuclear copper(II) complexes. These bis(macrocyclic) complexes catalyzed efficiently the selective oxidation of tetrahydrofuran into tetrahydrofuran-2-one and a small amount of tetrahydrofuran-2-ol and 4-hydroxybutyraldehyde using diluted H<sub>2</sub>O<sub>2</sub> as the oxidant. 1,1'-Diphenylsulfonbis(8,10-dimethyl-1,3,7,11,15-pentaazacyclohexadeca-7,11-diene)copper(II) perchlorate shows significantly higher catalytic activity than other bis(macrocycle) binuclear copper(II) complexes.



## Xiaolan Tang, Baocai Zhang, Yong Li, Yide Xu, The Pt-Sn/CeO<sub>2</sub> catalysts were found to be highly Qin Xin, Wenjie Shen Journal of Molecular Catalysis A: Chemical 235 (2005) 122

The role of Sn in Pt-Sn/CeO2 catalysts for the complete oxidation of ethanol

active for the complete oxidation of ethanol. The addition of tin resulted in the changes in both surface structure of metallic phase and the electronic Pt-Sn interaction, thus promoting the catalytic performances. The electronic configuration of Pt by Sn as well as the formation of PtSn alloy was the key factors in determining the nature of the active sites.



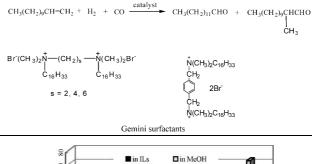
CH3

#### Min Li, Haiyan Fu, Min Yang, Hongjie Zheng, Yu-e He, Hua Chen, Xianjun Li

Journal of Molecular Catalysis A: Chemical 235 (2005) 130

Micellar effect of cationic gemini surfactants on organic/aqueous biphasic catalytic hydroformylation of 1-dodecene

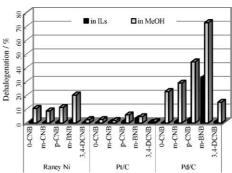
Hydroformylation of 1-dodecene catalyzed by water-soluble rhodium complex RhCl(CO)(TPPTS)<sub>2</sub> [TPPTS=tris(sodium-m-sulfonatophenyl) phosphine] was carried out in the micellar system. Higher reaction rates and higher regioselectivities toward the desired linear aldehyde were obtained in the cationic gemini surfactant solutions than those in the conventional monomeric surfactant CTAB solutions.



### Dan-Qian Xu, Zhi-Yan Hu, Wei-Wei Li, Shu-Ping Luo, Zhen-Yuan Xu

Journal of Molecular Catalysis A: Chemical 235 (2005) 137

Hydrogenation in ionic liquids: An alternative methodology toward highly selective catalysis of halonitrobenzenes to corresponding haloanilines



## Gao Zhenghong, Liu Zhongchen, He Fei, Xu Genhui

Journal of Molecular Catalysis A: Chemical 235 (2005) 143

Combined XPS and in situ DRIRS study of mechanism of Pd-Fe/α-Al<sub>2</sub>O<sub>3</sub> catalyzed CO coupling reaction to diethyl oxalate

The mechanism for the reaction of CO coupling to diethyl oxalate (DEO) over Pd-Fe/a-Al2O3 in gaseous phase at normal pressure was studied by X-ray photoelectron spectroscopy (XPS) and in situ diffuse reflectance infrared spectroscopy (DRIRS) techniques. The adsorption of two reactants, carbon monoxide and ethyl nitrite, on the catalyst was observed, respectively with in situ DRIRS as well as the catalysts after ethyl nitrite adsorption and after in situ reaction were also characterized by XPS measurement. These observational results show that (1) The ethyl nitrite on the catalyst surface has a dissociative chemisorption which leads to the oxidation of  $Pd^0$  to  $Pd^{2+1}$ (active component on the catalyst). So the reaction is a redox process; (2) Two intermediates, palladium complexes, in the reaction process are involved. From this, the mechanism steps of the reaction were proposed and the two intermediates were inferred.

 $Pd^{0} + 2RONO \rightarrow Pd^{0}(RONO),$  $\xrightarrow{2Pd^0} Pd^{2+}(^{-}OR)_2 + 2Pd - NO,$  $Pd^{2+}(^{-}OR)_2 + 2Pd^0 - CO \rightarrow Pd^{2+}(CO^{-}OR)_2$  $\rightarrow$  Pd<sup>0</sup> + ROOCCOOR

## Masoud Salavati-Niasari, Tahereh Khosousi, Samansa Hydarzadeh

Journal of Molecular Catalysis A: Chemical 235 (2005) 150

Highly selective esterification of *tert*-butanol by acetic acid anhydride over alumina-supported InCl<sub>3</sub>, GaCl<sub>3</sub>, FeCl<sub>3</sub>, ZnCl<sub>2</sub>, CuCl<sub>2</sub>, NiCl<sub>2</sub>, CoCl<sub>2</sub> and MnCl<sub>2</sub> catalysts

Esterification of *tert*-butanol by acetic anhydride to *tert*-butyl acetate over InCl<sub>3</sub>/Al<sub>2</sub>O<sub>3</sub>, GaCl<sub>3</sub>/Al<sub>2</sub>O<sub>3</sub>, FeCl<sub>3</sub>/Al<sub>2</sub>O<sub>3</sub>, MnCl<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub>, CoCl<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub>, NiCl<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub>, CuCl<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub>, and ZnCl<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub> catalysts have been investigated. The yield of *tert*-butyl acetate obtained was over 99.5%; when the reaction conditions used were as follows: the molar ratio of *tert*-butanol/acetic acid anhydride=1.1, the catalysts/(*tert*-butanol+acetic anhydride) wt. ratio=0.026, reaction temperature=30°C, and reaction time=2h. The esterification activity of the alumina supported metal chloride catalysts is in the following order: InCl<sub>3</sub>>GaCl<sub>3</sub>>FeCl<sub>3</sub>>ZnCl<sub>2</sub>>CuCl<sub>2</sub>>NiCl<sub>2</sub>>CoCl<sub>2</sub>>MnCl<sub>2</sub>.

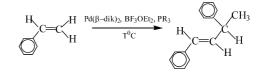
$$(CH_3)_3C-OH \frac{InCl_3-Al_2O_3, 30 \ ^{\circ}C}{Aceticanhydride}$$
  $(CH_3)_3C-OAc + CH_3COOH$ 

### Gomboo Myagmarsuren, Vitalii S. Tkach, Fedor K. Shmidt, Mesyef Mohamad, Dmitrii S. Suslov

The selective dimerization of styrene to 1,3-diphenyl-1-butene over  $Pd(\beta-diketonate)_/BF_3OEt_2$  catalyst systems in both "phosphine-free" and "phosphine-modified" fashions has been investigated. Styrene dimers up to 95% consists of *trans*-1,3-diphenyl-1-butene. Reaction products were characterized with 1H, 13C NMR, IR, MS-GC spectroscopy and GC analysis. Palladium hydride complexes are likely to be catalytically active species.

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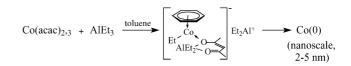
Selective dimerization of styrene to 1,3-diphenyl-1-butene with  $bis(\beta$ -diketonato)palladium/boron trifluoride etherate catalyst system



## F.K. Shmidt, L.O. Nindakova, B.A. Shainyan, V.V. Saraev, N.N. Chipanina, V.A. Umanetz

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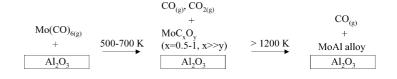
Hydrogenation catalysts formation in the system  $AlEt_3$ -Co(acac)<sub>2,3</sub>



## Y. Wang, F. Gao, W.T. Tysoe

Journal of Molecular Catalysis A: Chemical 235 (2005) 173

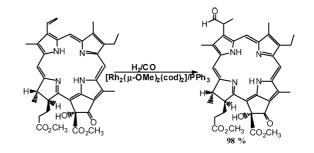
Interaction of molybdenum hexacarbonyl with dehydroxylated alumina thin films at high temperatures: Formation and removal of surface carbides



Andreia F. Peixoto, Mariette M. Pereira, Andreia F. Sousa, Alberto A.C. Pais, M. Graça P.M.S. Neves, Artur M.S. Silva, José A.S. Cavaleiro

Journal of Molecular Catalysis A: Chemical 235 (2005) 185

Improving regioselectivity in the rhodium catalyzed hydroformylation of protoporphyrin-IX and chlorophyll *a* derivatives

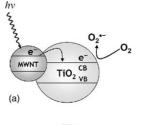


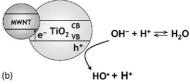
## Wendong Wang, Philippe Serp, Philippe Kalck, Joaquim Luís Faria

Journal of Molecular Catalysis A: Chemical 235 (2005) 194

Visible light photodegradation of phenol on MWNT-TiO<sub>2</sub> composite catalysts prepared by a modified sol-gel method

A synergetic effect could be ascribed to MWNT acting as photosensitizer in a photocatalytic process. This can be described in terms of (a) photo-induced electron transfer into the TiO<sub>2</sub> conduction band following light absorption by MWNT. The catalytic nature of the process demands that (b) back electron transfer occurs from the valence band of TiO<sub>2</sub> with formation of a hole, which is neutralized by oxidation of the adsorbed OH<sup>-</sup>.

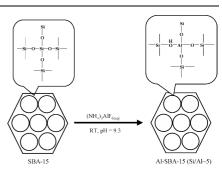




#### Hsien-Ming Kao, Chun-Chiang Ting, Shih-Wei Chao

Journal of Molecular Catalysis A: Chemical 235 (2005) 200

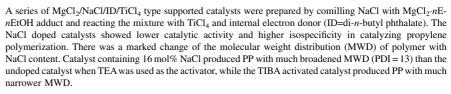
Post-synthesis alumination of mesoporous silica SBA-15 with high framework aluminum content using ammonium hexafluoroaluminate Incorporation of high aluminum contents into the silica framework of SBA-15 was achieved by treating SBA-15 with an aqueous  $(NH_4)_3AlF_6$  solution at pH 9.3 at room temperature. The Al-SBA-15 material thus obtained exhibited a high framework aluminum content (up to a bulk Si/Al ratio near 5), good structural integrity, and well-developed Brønsted acidity.

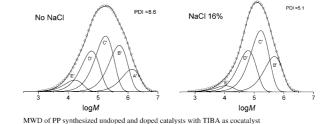


## Xue Jiang, Yong-ping Chen, Zhi-qiang Fan, Qi Wang, Zhi-sheng Fu, Jun-ting Xu

Journal of Molecular Catalysis A: Chemical 235 (2005) 209

 $\label{eq:prop} Propylene polymerization catalyzed by novel supported titanium catalysts MgCl_2/NaCl/DNBP/ TiCl_4 with different NaCl content$ 





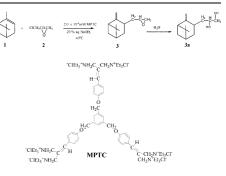
## E. Murugan, A. Siva

Journal of Molecular Catalysis A: Chemical 235 (2005) 220

Preparation of a novel soluble multi-site phase transfer catalyst and the kinetic study for the C-alkylation of  $\alpha$ -pinene

Novel soluble "multi-site (6 site)" phase transfer catalyst viz.1,3,5-tris(4-(2,2'-bis(N-triethylammoniummethylene chloride)eth-1-ene)phenoxymethyl)benzene (TBTEAPB) has been prepared by a simplified method and characterized. It is shown to be  $\cong$ 6 and 3 times as effective as singlesite and di-site PTCs respectively in catalyzing the C-alkylation of  $\alpha$ -pinene with epichlorohydrin.

Contents



#### D.J. Duvenhage, N.J. Coville

A series of Fe:Co/TiO<sub>2</sub> catalysts show lower activity than the equivalent Co/TiO<sub>2</sub> catalyst but with selectivity patterns similar to a Fe modified Co/TiO<sub>2</sub> catalyst.

(a) Precipitated(b) Incipient Wetness(c) MMCC

(d) MMC

700 800 900

**TPR** profiles

100 200 300

Relative Hydrogen uptake

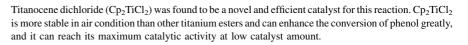
Journal of Molecular Catalysis A: Chemical 235 (2005) 230

 $Fe:Co/TiO_2$  bimetallic catalysts for the Fischer-Tropsch reaction. Part 4: A study of nitrate and carbonyl derived FT catalysts

## Hongying Niu, Jie Yao, Yue Wang, Gongying Wang

Journal of Molecular Catalysis A: Chemical 235 (2005) 240

Cp<sub>2</sub>TiCl<sub>2</sub> used as a catalyst for the transesterification between dimethyl carbonate and phenol to diphenyl carbonate



400 500 600

Temperature (°C)

$$2 - OH + CH_3O - C - OCH_3 \xrightarrow{Cp_2 TiCl_2} - O - C - O - C + 2CH_3OH$$

Anil Kumar, Sanjay S. Pawar

Henry reactions can be catalyzed in chloroaluminate ionic liquids containing excess of organic component with its easy recycling.

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Catalyzing Henry reactions in chloroaluminate ionic liquids

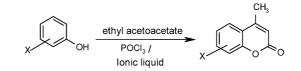
$$H$$
 + CH<sub>3</sub>NO<sub>2</sub>  $H$  + CH<sub>3</sub>NO<sub>2</sub>  $H$  NO<sub>2</sub>

## Mahesh K. Potdar, Meghana S. Rasalkar, Swapnil S. Mohile, Manikrao M. Salunkhe

Cleaner and economically viable protocols for coumarin synthesis via Pechmann condensation have been developed by employing neutral ionic liquids.

Journal of Molecular Catalysis A: Chemical 235 (2005) 249

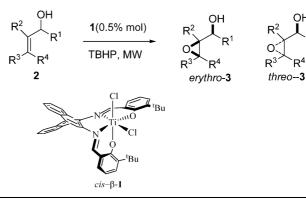
Convenient and efficient protocols for coumarin synthesis via Pechmann condensation in neutral ionic liquids



## Annunziata Soriente, Margherita De Rosa, Marina Lamberti, Consiglia Tedesco, Arrigo Scettri, Claudio Pellecchia

Journal of Molecular Catalysis A: Chemical 235 (2005) 253

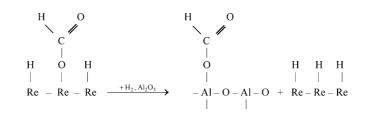
Synthesis, crystal structure and application in regio- and stereoselective epoxidation of allylic alcohols of a titanium binaphthyl-bridged Schiff base complex



## Frigyes Solymosi, Tímea Süli Zakar

Journal of Molecular Catalysis A: Chemical 235 (2005) 260

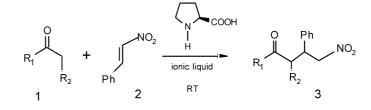
FT-IR study on the interaction of  $CO_2$  with  $H_2$  and hydrocarbons over supported Re



# Meghana S. Rasalkar, Mahesh K. Potdar, Swapnil S. Mohile, Manikrao M. Salunkhe

Journal of Molecular Catalysis A: Chemical 235 (2005) 267

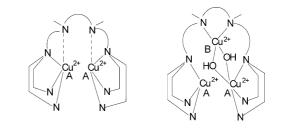
An ionic liquid influenced L-proline catalysed asymmetric Michael addition of ketones to nitrostyrene L-Proline catalysed Michael addition of different ketones to nitrostyrene is studied in ionic liquid. 1-Methoxyethyl-3-methylimidazolium methanesulfonate, ionic liquid, enhanced enantioselectivity of the reaction as compared to those carried out employing conventional organic solvents.



## Michele Gullotti, Laura Santagostini, Roberto Pagliarin, Alessandro Granata, Luigi Casella

Journal of Molecular Catalysis A: Chemical 235 (2005) 271

Synthesis and characterization of new chiral octadentate nitrogen ligands and related copper(II) complexes as catalysts for stereoselective oxidation of catechols Three new octadentate nitrogen ligands were employed for the synthesis of dinuclear and trinuclear copper(II) complexes. The complexes are active in the biomimetic oxidation of catechols to quinones and show stereoselective effects towards the optically active catechols l-Dopa and d-Dopa, and their methyl esters. In all cases, the preferred enantiomeric substrate has the l configuration.



## Elisabete da Palma Carreiro, Guo Yong-En, Anthony J. Burke

In this paper, we report our results on the evaluation of six chiral 2-substituted pyridine ligands in the MTO catalysed epoxidation of olefins.

Journal of Molecular Catalysis A: Chemical 235 (2005) 285

Approaches towards catalytic asymmetric epoxidations with methyltrioxorhenium(VII) (MTO): Synthesis and evaluation of chiral non-racemic 2-substituted pyridines

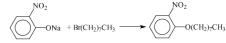
MTO (0.5 mol%)

#### Piao-Jen Lin, Hung-Ming Yang

Third-liquid phase-transfer catalyzed etherification for synthesizing o-nitrophenyl octyl ether:

Journal of Molecular Catalysis A: Chemical 235 (2005) 293

Kinetics for etherification of sodium *o*-nitrophenoxide via third-liquid phase-transfer catalysis



Aqueous reactant: sodium o-nitrophenoxide (ArONa) Organic reactant: 1-bromooctane (RBr) Phase-transfer catalyst: tetra-n-butylphosphonium bromide, etc. (QBr) Product: o-nitrophenyl octyl ether (ArOR)

Reaction mechanism:

Organic phase	ArOR + QBr <del>&lt;</del>	ArOQ + RBr
Third liquid phase	ArOR + QBr 🚤	
1 nira iiquia pnase	ArONa + QBr	
	11	11
Aqueous phase	ArONa + QBr	ArOQ + NaBr

## Jixiang Chen, Rijie Wang, Jiyan Zhang, Fei He, Sen Han

Journal of Molecular Catalysis A: Chemical 235 (2005) 302

Effects of preparation methods on properties of Ni/  $CeO_2\mbox{-}Al_2O_3$  catalysts for methane reforming with carbon dioxide

Nickel-based xerogel and aerogel catalysts have been prepared, and the method for the preparation of the aerogel leads to a higher specific surface area and higher dispersivity and homogeneity of nickel species than that of the xerogel.

